

Hong-Tao Fan, Markus  
Schürmann, Hans Preut\* and  
Norbert KrauseFachbereich Chemie, Universität Dortmund,  
Otto-Hahn-Strasse 6, 44221 Dortmund,  
GermanyCorrespondence e-mail:  
uch002@uxp1.hrz.uni-dortmund.de

## Key indicators

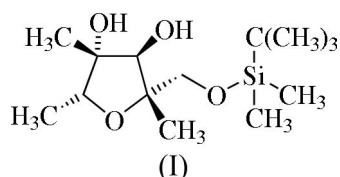
Single-crystal X-ray study  
 $T = 173$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.012$  Å  
 $R$  factor = 0.067  
 $wR$  factor = 0.182  
Data-to-parameter ratio = 8.8For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.**(2*S*,3*R*,4*R*,5*R*)-2-(*tert*-Butyldimethylsilyloxymethyl)-  
2,4,5-trimethyltetrahydrofuran-3,4-diol**

In the title compound,  $\text{C}_{14}\text{H}_{30}\text{O}_4\text{Si}$ , the bond angles around Si are in the range  $105.4$  (3)– $112.1$  (4) $^\circ$  and the tetrahydrofuran ring adopts a distorted envelope conformation. There is one intramolecular and one intermolecular  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bond in the crystal structure. A spiral of molecules forms around the  $4_1$  axis ( $c$  axis) of the crystal structure *via* the intermolecular  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bond.

Received 17 May 2005  
Accepted 26 May 2005  
Online 10 June 2005

## Comment

The title compound, (I), is an intermediate in the enantioselective synthesis of the natural products citreoviral and citreoviridin. The compound was obtained in a multistep sequence involving a gold-catalysed cycloisomerization (Hoffmann-Röder & Krause, 2001). The crystal structure proves the relative configuration of the four stereogenic centres within the tetrahydrofuran (THF) ring. The five-membered ring has the shape of a slightly distorted envelope with the torsion angles  $31.8$  (8),  $-44.5$  (7),  $40.3$  (8),  $-18.9$  (8) and  $-9.6$  (8) $^\circ$ .



## Experimental

The synthesis of (I) will be described elsewhere (Fan & Krause, 2005). It was dissolved in a small amount of THF and hexane, and crystals were obtained by vapour diffusion of hexane.

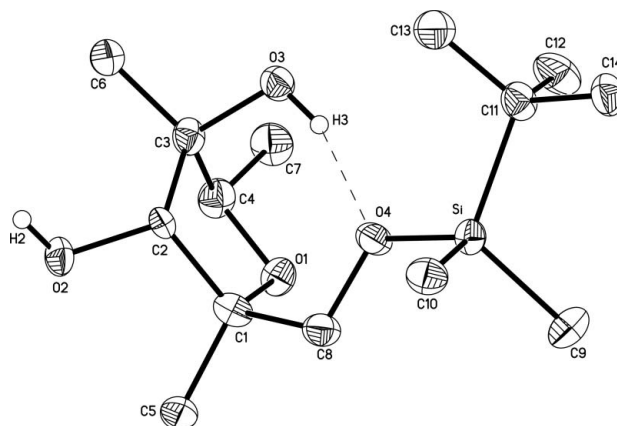


Figure 1

The molecular structure of the title compound, showing the labelling of all non-H atoms and of the H atoms involved in  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds. The remaining H atoms have been omitted for clarity. The dashed line indicates the intramolecular hydrogen bond. Displacement ellipsoids are shown at the 30% probability level.

Crystal data

C<sub>14</sub>H<sub>30</sub>O<sub>4</sub>Si  
*M<sub>r</sub>* = 290.47  
 Tetragonal, *P*4<sub>1</sub>  
*a* = 12.3616 (12) Å  
*c* = 11.1685 (11) Å  
*V* = 1706.6 (3) Å<sup>3</sup>  
*Z* = 4  
*D<sub>x</sub>* = 1.130 Mg m<sup>-3</sup>

Mo *K*α radiation  
 Cell parameters from 20589 reflections  
 $\theta = 3.0\text{--}25.4^\circ$   
 $\mu = 0.14 \text{ mm}^{-1}$   
*T* = 173 (1) K  
 Needle, colourless  
 0.36 × 0.08 × 0.08 mm

Data collection

Nonius KappaCCD diffractometer  
 $\omega$  scans  
 Absorption correction: none  
 20589 measured reflections  
 1607 independent reflections  
 1227 reflections with *I* > 2σ(*I*)

*R*<sub>int</sub> = 0.049  
 $\theta_{\text{max}} = 25.4^\circ$   
*h* = -14 → 14  
*k* = -10 → 10  
*l* = -13 → 13

Refinement

Refinement on *F*<sup>2</sup>  
*R* [*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.067  
*wR* (*F*<sup>2</sup>) = 0.182  
*S* = 1.15  
 1607 reflections  
 182 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0495P)^2 + 4.0462P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.27 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.26 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O2—H2...O3 <sup>1</sup>	0.82	2.02	2.834 (7)	170
O3—H3...O4	0.82	1.97	2.750 (8)	158

Symmetry code: (i) -*y* + 1, *x*, *z* +  $\frac{1}{4}$ .

In the absence of significant anomalous scattering effects, Friedel pairs were merged. H atoms were placed in calculated positions, with C—H = 0.96–0.98 Å and O—H = 0.82 Å, and were refined as riding, with *U*<sub>iso</sub>(H) values of 1.5*U*<sub>eq</sub>(C,O) for methyl and hydroxy groups, and 1.2*U*<sub>eq</sub>(C) for others; the methyl and hydroxy groups were allowed to rotate but not to tip.

Data collection: COLLECT (Nonius, 1998); cell refinement: DENZO and SCALEPACK (Otwinowski & Minor, 1997); data reduction: DENZO and SCALEPACK; program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL-Plus (Sheldrick, 1991); software used to prepare material for publication: SHELXL97, PARST95 (Nardelli, 1995) and PLATON (Spek, 2003).

References

Fan, H.-T. & Krause, N. (2005). *Org. Lett.* In preparation.  
 Hoffmann-Röder, A. & Krause, N. (2001). *Org. Lett.* **3**, 2537–2538.  
 Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.  
 Nonius (1998). COLLECT. Nonius BV, Delft, The Netherlands.  
 Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326, New York: Academic Press.  
 Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.  
 Sheldrick, G. M. (1991). SHELXTL-Plus. Release 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.  
 Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.  
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.